PICRASINOL D, A NEW QUASSINOID FROM THE STEM WOOD OF PICRASMA AILANTHOIDES

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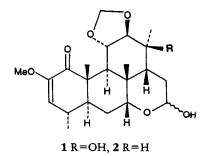
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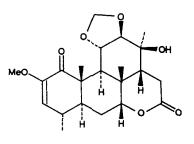
ABSTRACT.—A new quassinoid, picrasinol D [1], was isolated from the stem wood of *Picrasma ailanthoides*. The structure was elucidated by spectral evidence and chemical transformation.

More than 20 quassinoids (aglyconeš) have been isolated by Murae et al. (1-6)and Hikino et al. (7) from Picrasma ailanthoides Planchon (Simaroubaceae). We have isolated eight new quassinoid glucosides, picrasinosides A–H, and three new quassinoid hemiacetals, picrasinols A, B, and C, from the stem bark of P. ailanthoides (8–10). Further study on the stem wood of this plant has led to the isolation of a new quassinoid, picrasinol D [1]. This report deals with the isolation and structure elucidation of this compound.

The new compound [1] was obtained as an amorphous solid. Its ir spectrum showed the presence of hydroxyl (3450 cm⁻¹) and α,β -unsaturated carbonyl (1695 cm⁻¹) groups. The uv spectrum of 1 exhibited maximal absorption at 256 nm due to a conjugated enone system. Compound 1 was inferred as being a new quassinoid, because its ir and ¹H-nmr spectra did not coincide with those of any of the known quassinoids, and it was named picrasinol D. Picrasinol D [1] had the molecular formula $C_{22}H_{32}O_7$ as determined by hreims. The ¹³C-nmr signals (Table 1) observed for C-1 to C-12, C-14, C-15, and the methylenedioxy moiety in **1** were nearly identical with those of **2**(8). However, the ¹³C-nmr signals for C-13 and Me-13 of **1** were found at lower field than those of **2**. Chemical shift values (ppm) for the C-13 and Me-13 signals were 40.8 and 11.1, respectively. The DEPT spectra of **1** and **2** showed that the C-13 of **1** is a quaternary carbon, whereas C-13 of **2** is a methine carbon.

The ¹H-nmr signals of **1** (Table 2) were similar to those of **2** except for those of Me-8 and Me-13. The Me-8 and Me-13 signals of **1** appeared at lower field by 0.21 and 0.46 ppm each, respectively, compared to analogous signals for **2**. On the other hand, the ¹H- and ¹³C-nmr signals of **1** coincided with those of **3** (11), except for the signals at H-16 and C-16, as shown in Tables 1 and 2. These findings, therefore, suggested that **1** has a hydroxyl group at C-13. Additionally,





Carbon	Compound		
	1	2	3
C-1	199.0 (C=O)	198.7 (C=O)	198.1 (C=O)
C-2	149.3 (C)	148.2 (C)	149.0 (C)
OMe-2	54.8 (CH ₃)	54.8 (CH ₃)	54.9 (CH ₃)
C-3	115.2 (CH)	115.3 (C)	115.5 (C)
C-4	32.3 (CH)	32.2 (CH)	32.0 (CH)
Me-4	19.4 (CH ₃)	19.4 (CH ₃)	19.2 (CH ₃)
C-5	44.5 (CH)	44.3 (CH)	43.4 (CH)
C-6	26.3 (CH ₂)	26.4 (CH ₂)	25.8 (CH ₂)
C- 7	78.3 (CH)	79.1 (CH)	81.7 (CH)
C-8	40.0 (C)	39.0 (C)	38.7 (C)
С-9	38.6 (CH)	37.2 (CH)	38.2 (CH)
C-10	47.6 (C)	47.4 (C)	47.2 (C)
Me-10	13.4 (CH ₃)	13.1 (CH ₃)	13.3 (CH ₃)
C-11	76.1 (CH)	78.3 (CH)	75.4 (CH)
C-12	85.4 (CH)	83.3 (CH)	84.4 (CH)
C-13	74.3 (C)	33.5 (C)	73.9 (C)
Me-13	26.1 (CH ₃)	15.0 (CH ₃)	24.0 (CH ₃)
C-14	54.3 (CH)	48.5 (CH)	51.3 (CH)
C-15	31.3 (CH ₂)	31.1 (CH ₂)	31.0 (CH ₂)
C-16	91.4 (CH)	93.1 (CH)	170.4 (C=O)
-OCH ₂ O	95.7 (CH ₂)	95.5 (CH ₂)	95.9 (CH ₂)

TABLE 1. ¹³C-Nmr Spectra of Compounds 1–3.^a

^aValues are in ppm. Measured at 125.7 MHz in C₅D₅N.

the molecular formula of 1 obtained from hreims indicated that 1 has one more oxygen atom than 2.

Oxidation of 1 with Jones reagent in Me₂CO afforded a lactone compound whose ir and ¹H-nmr spectral data and hplc retention time were identical with those of 3. From an analysis of all of the above data, the structure of the isolate from *P. ailanthoides* was proposed as 1.

The structure of 1 was also confirmed by CH long-range correlations in the HMBC nmr spectrum, which showed the following correlations: the methyl protons at δ 3.49 (OMe-2) with the carbon at δ 149.3 (C-2); the methyl protons at δ 0.99 (Me-4) with the carbons at δ 32.3 (C-4) and δ 44.5 (C-5); the olefinic proton at δ 5.20 (H-3) with the carbons at δ 199.0 (C-1) and δ 44.5 (C-5); the methyl protons at δ 1.76 (Me-8) with the carbon at δ 78.3 (C-7); the methyl protons at δ 1.51 (Me-13) with the carbon at δ 54.3 (C-14); the methine proton at δ 3.30 (H-9) with the carbons at δ 199.0 (C-1), δ 40.0 (C-8), δ 76.1 (C-11), and δ

85.4 (C-12); the methine proton at δ 2.53 (H-14) with the carbons at δ 40.0 (C-8) and δ 31.3 (C-15); the methyl protons at δ 1.54 (Me-10) with the carbons at δ 199.0 (C-1) and δ 47.6 (C-10); and the methine proton at δ 3.86 (H-12) with the carbon at δ 76.1 (C-11).

The relative stereochemistry of **1** was confirmed by nOe correlations that were observed between Me-4 and H-4, Me-4 and H-5, Me-8 and H-7, Me-8 and H-14, Me-10 and H-11, Me-10 and H-4, Me-10 and Me-8, H-7 and H-14, H-9 and H-12, and H-12 and Me-13. Thus, the relative stereochemistry of **1** was determined as shown in Figure 1.

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.—Mps were determined on an MRK air bath-type melting point apparatus and were uncorrected. Specific rotations were observed on a Jasco DIP-370 polarimeter (cell length=0.1 dm). Ir and uv spectra were recorded on a Jasco IR-810 spectrometer and a Hitachi 320-S spectrometer, respectively. Nmr spectra were recorded on a Varian URX-500 spectrometer (499.8 MHz for ¹H and 127.5 MHz for

Proton	Compound		
	1 ^b	2 ^b	3 °
OMe-2 H-3 H-4 Me-4 H-5 H-7 Me-8 H-9 Me-10 H-11 H-12 Me-13 H-14 H-16	3.49 s 5.20 d (2) 2.31 m 0.99 d (6.5) 2.20 m 3.44 br s 1.76 s 3.30 d (11.5) 1.54 s 4.44 dd (11.5,9) 3.86 d (9) 1.51 s 2.53 t (9.5) 5.66 br s	3.48 s 5.21 d (2) 2.15 m 0.93 d (6.5) d 3.37 br s 1.09 s 3.19 d (11) 1.42 s 3.66 dd (11,10) 3.56 d (10) 1.05 d (6.5) d d	3.49 s 5.20 d (2.2) 2.18 m 0.88 d (6.8) 2.20 br s 4.27 br s 1.72 s 2.99 d (11) 1.50 s 4.20 dd (11,10) 3.69 d (10) 1.49 s 2.12 t (11)
-OCH ₂ O	5.15 br s 5.36 br s	5.11 s 5.26 s	5.18 br s 5.39 br s

TABLE 2. ¹H-Nmr Spectra of Compounds 1-3.^{*}

 $^{a}Values$ are δ ppm. Coupling constants in Hz in parentheses.

^bMeasured at 500 MHz in C₅D₅N.

^cMeasured at 200 MHz in C₅D₅N.

^dNot assignable.

 13 C) and a Varian XL-200 spectrometer (200 MHz for ¹H) in C₅N₅N, using TMS as an internal standard. Mass spectra (eims, hreims) were recorded on a Hitachi M-80 spectrometer. Si gel (Merck, type 60, 70–230 mesh) was used for cc. Precoated Si gel plates (Merck 60 F₂₃₄ of 0.25-mm thickness) were used for analytical tlc and plates of 1-mm and 2-mm thickness were used for prep. tlc. Analytical hplc was performed on a Tosoh liquid chromatograph equipped with a detector at 245 nm and a reversed-phase column (TSK-gel ODS-80T_M), using a mixed solvent of MeOH/H₂O. Prep. hplc was carried out with the same solvent system using reversed-phase columns (Dynamax-60A and Lichrosorb RP-60).

PLANT MATERIAL.—The stems of *P. ailanthoides* were procured and identified by Professor K. Kondo, Laboratory of Plant Chromosome and Gene Stock, Faculty of Science, Hiroshima University. A voucher specimen (No. 8201) has been deposited in the Department of Interdisciplinary Studies of Natural Environment, Hiroshima University.

EXTRACTION AND ISOLATION.—Half-dried stems of *P. ailanthoides* were separated into bark and wood. The wood (40 kg) was cut into small chips and soaked in MeOH (90 liters) at room temperature for two weeks. The fatty substances in the MeOH extract were removed by extraction

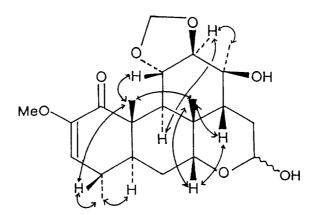


FIGURE 1. NOe correlations for 1.

with hexane $(2\times)$, and the material used for this investigation was obtained by CHCl₃ extraction $(3\times)$ as a brown resinous substance (143 g) (9). The resinous substance (100 g) was chromatographed on a Si gel column eluted with a mixed solvent of CHCl₃-MeOH-H₂O (50:14:3, lower layer) and MeOH to afford 31 fractions. Tlc and hplc analysis of all these fractions indicated that fractions 5–11 contained quassinoid compounds.

Fraction 11 (1.91 g) was subjected to prep. hplc [Dynamax-60A, MeOH-H₂O(1:1)] to afford 10 fractions. The fifth fraction contained quassinoid 1 as the major component. Careful repeated prep. hplc [Lichrosorb RP-18, MeOH-H₂O(1:1)] of the fraction led to the isolation of picrasinol D [1] (49.3 mg, 0.00017%).

Picrasinol D[1].—Colorless amorphous solid: mp 138–140°; $[\alpha]^{21}D + 38.2°$ (*c*=0.17, EtOH); uv λ max 255 (€ 4140) nm; ir (KBr) ν max 3450 (OH), 1695 (α,β-unsaturated C=O), 1625 (C=C) cm⁻¹; ¹³C-nmr data, see Table 1; ¹H-nmr data, see Table 2; hreims *m*/z found [M]⁺ 408.2154 (calcd for C₂₂H₃₂O₇, 408.2147).

Oxidation of picrasinol D [1].—Picrasinol D [1] (5.0 mg, 0.012 mM) was dissolved in Me₂CO (10 ml). Jones reagent (1 ml, 2.7 mM) in Me₂CO (10 ml) was added dropwise to the solution stirring at 0°. The solution was stirred at 25° for 2 h, after which EtOH (0.2 ml) and H₂O (10 ml) were added. The resulting solution was extracted with CHCl₃ (5 ml) three times. The CHCl₃ solution was washed with H₂O, dried over MgSO₄, and evaporated. Purification of the crude product by prep. hplc [Lichrosorb RP-18, MeOH-H₂O (1:1)] afforded compound **3** (3.6 mg, 72% yield).

Compound 3.—Colorless amorphous solid: mp 260–262°; ir (KBr) ν max 3375 (OH), 1710 (δ -lactone C=O), 1705 (α , β -unsaturated C=O), 1630 (C=C) cm⁻¹; ¹H nmr (200 MHz, C₅D₅N) δ 0.88 (3H, d, J=7.0 Hz, Me-4), 1.49 (6H, s, Me10 and Me-13), 1.72 (3H, s), 3.49 (3H, s, OMe-2), 5.20 (1H, d, J=2.0 Hz, H-3), 5.18 (1H, s, -OCH₂O-), 5.39 (1H, s, -OCH₂O-); eims m/z[M]⁺ 406 (100).

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LITERATURE CITED

- T. Murae, T. Tsuyuki, T. Ikeda, T. Nishihama, S. Masuda, and T. Takahashi, *Tetrabedron*, 27, 1545 (1971).
- T. Murae, T. Tsuyuki, T. Ikeda, T. Nishihama, S. Masuda, and T. Takahashi, *Tetrabedron*, 27, 5147 (1971).
- T. Murae, A. Sugie, T. Tsuyuki, S. Masuda, and T. Takahashi, *Tetrahedron*, 29, 1515 (1973).
- T. Murae, T. Ikeda, A. Sugie, T. Nishihama, T. Tsuyuki, and T. Takahashi, Bull. Chem. Sac., 46, 3621 (1973).
- T. Murae, A. Sugie, Y. Moriyama, T. Tsuyuki, and T. Takahashi, Org. Mass Spectrom., 8, 297 (1974).
- T. Murae, A. Sugie, T. Tsuyuki, and T. Takahashi, *Chem. Pharm. Bull.*, 23, 2188 (1975).
- H. Hikino, T. Ohta, and T. Takemoto, *Phytochemistry*, 14, 2473 (1974).
- M. Okano, T. Fujita, N. Fukamiya, and T. Aratani, Bull. Chem. Soc., 58, 1973 (1985).
- T. Matsuzaki, N. Fukamiya, M. Okano, T. Fujita, T. Tagahara, and K.H. Lee, J. Nat. Prod., 54, 844 (1991).
- M. Daido, N. Fukamiya, M. Okano, and K. Tagahara, J. Nat. Prod., 55, 1643 (1992).
- T. Murae, A. Sugie, T. Tsuyuki, S. Masuda, and T. Takahashi, *Tetrabedron*, **29**, 1515 (1973).

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